Investigations in the Field of Chemical Transformations SOV/62-59-1-25/38 of Unsaturated and High-Molecular Compounds. Communication 8. - Copolymerization of γ -Silicon-Containing Vinyl Ethers and Methyl Methacrylate

butyric acid. According to the experimental data obtained the following regularities were found: on the increase of y-silicon-containing vinyl ether in the reaction medium the yield of copolymers is decreased while the number of the members of vinyl ether in them is increased (Fig). Similar rules have been already observed in the copolymerization of vinyl ether and vinyl ester (Ref 3). As may be seen from it, the content of γ -silicon-containing vinyl ether in the copolymer does not exceed 50 mol-%. The polymerization according to radical mechanism was not observed with ysilicon-containing vinyl ether. As already mentioned in reference 4, it may be assumed that in this case reaction is started by a complex radical. The latter is produced by the addition of the more active monomer of methyl methacrylate to the radical which was formed in the decomposition of the initiator. A comparison between y-silicon-containing vinyl ether and the vinyl alkyl ethers demonstrated that the reactivity of vinyl ether is reduced by the presence of silicon in y-position (Table 1). The results of investiga-

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Investigations in the Field of Chemical Transformations SOV/62-59-1-23/38 of Unsaturated and High-Molecular Compounds. Communication 8. ... Copolymerization of γ -Silicon-Containing Vinyl Ethers and Methyl Methacrylate

tion mentioned in stable 2 permit the conclusion that the substitution of ethyl groups for methyl groups reduces somewhat the yield of copolymers in the case of \gamma-silicon-containing ether. However, the composition of the copolymers is hardly affected by that. In the investigation of the copolymerization of \gamma-silicon-containing vinyl ether and methyl methacrylate it was stated that their copolymers receive new properties in the presence of silicon. There are 1 figure, 3 tables and 4 Soviet references.

ASSOCIATION:

Institut organicheskoy khimii im. N. D. Zelinskogo Akademii nauk SSSR (Institute of Organic Chemistry imeni N. D. Ze-

linskiy of the Academy of Sciences, USSE)

SUBMITTED:

May 22, 1957

Card 3/3

SHIKHIYEV, I.A.; SHOSTAKOVSKIY, M.F.; KAYUTENKO, L.A.

Investigations in the field of the synthesis and transformation of unsaturated silicon organic compounds. Dokl.AN Azerb.SSR 15 no.1:21-23 (MIRA 12:3)

1. Institut organicheskoy khimii AN SSSR i Institut nefti AN AzerSSR. Predstavleno akademikom AN AzerSSR Yu.G. Mamedaliyevym. (Silicon organic compounds)

SHIKHIYEV, I.A.; ALIYEV, M.I.; KARAYEVA, Sh.V.

Synthesis and conversion of tertiary y -acetylenic alcohols containing silicon. Dokl.AN Azerb.SSR 15 no.12:1111-1113 159. (MIRA 13:4)

1. Institut neftekhimicheskikh protsessov AN AzerSSR. Predstavleno akademikom AN AzerSSR M.F. Nagiyevym.
(Alcohols)

5 (3)

AUTHORS: Shostakovskiy, M. F., Komarov, N. V.

SOV/74-28-6-4/5

Shikhiyev, I. A.

CONCRETE SHOWING PROPERTY TO THE THEORY

TITLE:

Silanols (Silanoly)

PERIODICAL:

Uspekhi khimii, 1959, Vol 28, Nr 6, pp 741 - 771 (USSR)

ABSTRACT:

The present paper describes the methods of synthesis and the properties of silanols. Silanols are compounds containing hydroxyl groups directly bound to the silicon atom. Owing to particular affinity of the silicon atom to oxygen, there are silicon-organic compounds with one, two and three hydroxyl groups with the silicon atom. In this connection, they are classified into alkyl (aryl) silanols with the common formula R₂SiOH, silandiols R₂Si(OH)₂, and silantriols RSi(OH)₃. Of the mentioned compounds, the trialkyl (aryl) silanols are best investigated. They are highly reactive compounds and have great practical and theoretical importance. In contrast to silicon-organic alcohols, there are a number of common methods of synthesis for silanols. For some representatives of this class, there are quitespecial methods of synthesis peculiar to them. Of the methods of synthesis, the following were described:

Card 1/4

Silanols

SOV/74-28-6-4/5

the hydrolysis of halogen silanes (Refs 8, 10-63), of alcoxysilanes (Refs 25, 35, 64-75), of acetoxysilanes (Refs 10, 76-81), of aminosilanes (Refs 10,21,82-90), the magnesium-organic synthesis (Refs 91-95), the cleavage of siloxanes (Refs 95-103), the cleavage of tetrasubstituted silanes (Refs 102,104-124), and the hydrolysis of hydride silanes (Refs 35,60,84,128-132). Besides the mentioned general methods of obtaining silanols, special publications describe many other methods which are suitable for the synthesis of compounds with a certain structure (Refs 5,24,31,102,103,111,124,133-147). The physical properties of many silanols have not yet been fully characterized (Table). For some representatives, the physical constants are contradictory. In spite of this, certain rules referring to their physico-chemical properties can be determined from the data indicated in the table of the compounds of this class (Refs 32,148-155). Trialkyl (aryl) silanols remind of tertiary alcohols as to their structure. Silandiols and silantriols have no analoga among organic compounds. Silanols are, in general, similar to the corresponding organic alcohols, but by the presence of the silicon atom their properties attain a character peculiar to them. This brings about that in many ca-

Card 2/4

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Silanols

SOV/74-28-6-4/5

ses they react differently from alcohols. Some of these reactions were considered closely and compared with the corresponding reactions of organic analoga: action of metals and lyes (Refs 8,12,31,43,60,68,81,96-103,203,205-222), dehydration of silanols (Refs 4,24,26,33,35,37,46,47,60,63,68,190,224-226), interaction with halogen silanes (Refs 11,21,128,227,230,231), action of mineral acids (Refs 31,62,228,232,233), interaction with acid anhydrides and halogen anhydrides (Refs 22,31,41,61, 223,227,324,235), interaction with alcoxysilanes (Refs 21,230, 236-239), reaction with vinyl ether (Refs 52,80,87-90,226,240-244), action of Mischer's reagent (Refs 169,245), interaction with isocyanates (Refs 37,51,183,246), hydration of silanols (Refs 37,247). As mentioned before, silanols are highly reactive substances which are capable of undergoing various reactions. Some of these reactions were described (Refs 37,21,49,53, 81,102,139,167,169,172,206-209,223,249-264).Silanols are used in industry for the production of various resins (Refs 257-265), heat-resisting coats (Refs 203,209), bactericide substances (Ref 265), adhesives (Refs 267,268), water-repellent agents (Refs 210, 268), for impregnating paper (Ref 270), as diffusion

Card 3/4

Silanols

SOV/74-28-6-4/5

liquids (Ref 237), for lubricating oils (Ref 237) and for many other substances. There are 1 table and 270 references, 65 of

which are Soviet.

ASSOCIATION:

In-t organicheskoy khimii im. N. D. Zelinskogo AH SSSR (Institute of Organic Chemistry imeni N. D. Zelinskiy, AS USSR)

Card 4/4

AUTHORS:

Shostakovskiy, M. F., Shikhiyev, I. A., SOV/79-29-2-5/71

Komarov, N. V.

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TITLE:

Synthesis and Transformations of Organosilicon Vinyl Ethers (Sintez i prevrashcheniya kremneorganicheskikh prostykh vinilo-

vykh efirov)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 2, pp 366-376 (USSR)

ABSTRACT:

Contrarily to vinyl ethers (Ref 1) the synthesis and the properties of organosilicon vinyl ethers are not mentioned in publications at all. The present paper describes the vinyl ethers of γ -silicon-containing alcohols according to Favorskiy-Shostakovskiy. At the same time it must be observed that the synthesis of the silanol ethers of the α - and β -silicon-containing alcohols did not succeed in this way. The common vinylating method, using caustic potash as well as Na and potassium silanolates, proved inadequate for the purpose. The use of alcoholates of the same reacting alcohols favors vinylation. According to the method worked out, the synthesis of the γ -silicon-containing vinyl ethers was carried out:

Card 1/2

 $^{2R}3^{\text{SiCH}}2^{\text{OH}} + \text{CH} \equiv \text{CH} \longrightarrow ^{R}3^{\text{SiCH}}2^{\text{O-CH}}2^{\text{SiR}}3^{\text{H}}2^{\text{O}}$, where $^{R\Xi}$ alkyl.

Synthesis and Transformations of Organosilicon Vinyl Ethers

SOV/79-29-2-5/71

The affiliation of alcohols, silanols, and organic acids to the γ -silicon-containing vinyl ethers takes place under the influence of acid catalysts according to the ion mechanism, in agreement with Markovnikov's rule. On their reaction with organosilicon alcohols and silanols, hitherto unknown acetals were obtained, containing silicon atoms in both alcohol radicals. The reaction of these ethers with organic acids can serve as a basis for the synthesis of a new class of organosilicon compounds, i.e. organosilicon acylates. Hydrogenation, chlorination, and hydrochlorination of the γ -silicon-containing vinyl ethers were investigated. There are 3 tables and 12 Soviet references.

ASSOCIATION:

Institut organicheskoy khimii Akademii nauk SSSR (Institute for Organic Chemistry of the Academy of Sciences, USSR)

SUBMITTED:

November 5, 1957

Card 2/2

CIA-RDP86-00513R001549420004-2 "APPROVED FOR RELEASE: 08/25/2000

5 (3) AUTHORS: Shikhiyev, I. A., Shostakovskiy, M. F., SOV/79-29-5-31/75

Komarov, N. V. Aslanov, I. A.

CHICAGO BERNESE CONTRACTOR OF CONTRACTOR CON

TITLE:

Investigations in the Field of Synthesis and Transformations

of Unsaturated Organo-germanium Compounds (Issledovaniya v

oblasti sinteza i prevrashcheniy nepredel'nykh

germaniyorganicheskikh sovedineniy). I. Synthesis of Mono-, Di-, and Trisubstituted Tertiary γ -Germanium Acetylene Alcohols (Sintez odno-, dvukh- i trekhatomnykh tretichnykh

7-germaniyatsetilenovykh spirtov)

PERICDICAL:

Thurnal obshchey khimii, 1959, Vol 29, Nr 5,

pp 1549-1551 (USSR)

ABSTRACT:

In a previous paper (Ref 1) the reaction of di-magnesiumdibromodimethyl-ethinyl carbinol (Iotsich Roagent) with various alkyl-(aryl)-chlorosilanes was investigated. In order to explain further the reaction process with this

reagent, its influence upon methyl-, dimethyl-, and trimethyl germanium bromides was investigated. The reaction was found

to proceed with the formation of tertiary 7-germanium containing acetylene alcohols according to the following

Cord 1/3 scheme:

Investigations in the Field of Synthesis and 50V/79-29-5-31/75 Transformations of Unsaturated Organo-germanium Compounds. I. Synthesis of Mono-, Di-, and Trisubstituted Tertiary &-Germanium Acetylene Alcohols

The occurrence of hydroxyl groups was confirmed by the acetic acid derivative of bis-(2-methylbutin-3-ol-2)-dimethyl germanium. Further reactions of organo-germanium elcohols and their derivatives will be described in later papers. The experimental part presents the physical data of the initial substances, the details of the synthesis, and the analysis of the compounds obtained. The authors prepared: (2-methyl-butin-3-ol-2)-4-trimethyl germanium, bis-(2-methyl-butin-3-ol-2)-4-dimethyl germanium, tri-(2-methyl-butin-3-ol-2)-4-methyl germanium, and bis-(2-methyl-butine-3-acetoxy-2)-4-dimethyl

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Investigations in the Mield of Synthesis and SOV/79-29-5-31/75 Transformations of Unsaturated Organo-germanium Compounds. I. Synthesis of Mono-, Di-, and Trisubstituted Tertiary 7-Germanium Acetylene Alcohols

germanium. There are 3 Soviet references.

SUBLATTED:

Hovember 25, 1957

Card 3/3

5(3) AUTHORS: SOV/79-29-7-7/83

Shikhiyev, I. A., Shostakovskiy, M. F., Kayutenko, L. A.

TITLE:

Investigations in the Field of the Synthesis and the Transformations of Unsaturated Organo-silicon Compounds (Issledovaniya v oblasti sinteza i prevrashcheniy nepredel'nykh kremneorga-

nicheskikh soyedineniy).

II. Synthesis of the Silicon Hydrocarbons of the Vinyl Acetylene Series (II. Sintez kremneuglevodcrodov vinilatsetilenovogo ryada)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 7, pp 2137-2139 (USSR)

ABSTRACT:

The synthesis of acetylene alcohols (Ref 1), their hydrogenation and dehydration (Refs 2, 3) as well as the affiliation of various compounds to the triple bond (Refs 1, 4, 5) is of high theoretical and practical interest. Similar conversions of the organo-silicon acetylene alcohols were carried out for a comparative investigation of their properties. Earlier, the authors elaborated the synthesis of mono- (Ref 6), ti- (Ref 7), and trivalent (Ref 8) %-silicon and %-germanium substituted (Ref 9) acetylene alcohols. In the present paper the dehydration conditions of some %-silicon substituted ditertiary acetylene

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glycols as well as the catalytic hydrogenation of the

Investigations in the Field of the Synthesis and the SOV/79-29-7-7/83 Transformations of Unsaturated Organo-silicon Compounds. II. Synthesis of the Silicon Hydrocarbons of the Vinyl Acetylene Series

silicon hydrocarbons obtained were investigated according to the afore mentioned scheme. Thus, the synthesis of vinyl acetylene silicon hydrocarbons was elaborated by the dehydration of the corresponding ditertiary y-silicon substituted acetylene glycols in the presence of KHSO₄. The following compounds were obtained and characterized: bis-(2-methyl-butene-1-in-3)-ethyl silane; bis-(2-methyl butene-1-in-3)-diethyl silane; bis-(2-methyl butene-1-in-3)-dimethyl silane; bis-(2-methyl butene-1-in-3)-methyl silane, and bis-(2-methyl butene-1-in-3)-methyl propyl silane. By catalytic hydrogenation of bis-(2-methyl butene-1-in-3)-diethyl silane the corresponding saturated silicon hydrocarbon, diethyl diisoamyl silane, was

Card 2/3

Investigations in the Field of the Synthesis and the SOV/79-29-7-7/83 Transformations of Unsaturated Organo-silicon Compounds. II. Synthesis of the Silicon Hydrocarbons of the Vinyl Acetylene Series

> synthesized. The silicon hydrocarbons synthesized are more exactly characterized in the table. There are 2 tables and 9 references, 8 of which are Soviet.

ASSOCIATION:

Institut organicheskoy khimii Akademii nauk SSSR i Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Organic Chemistry of the Academy of Sciences USSR and Institute of Petrochemical Processes of the Academy of

Sciences of the Azerbaydzhanskaya SSR)

SUBMITTED:

July 3, 1958

医马克里氏试验 医阿拉克氏试验检尿病 医抗性性 医动物性 医神经性性 医神经性神经神经神经神经

Card 3/3

PHASE I BOOK EXPLOITATION

SOV/4989

Shikhiyev, Ibragim Abasovich, Professor, Doctor of Chemical Sciences, Mikhail Fedorovich Shostakovskiy, Professor, Doctor of Chemical Sciences, Nikolay Vasil'yevich Komarov, Candidate of Chemical Sciences

- Novyye, kislorodsoderzhashchiye kremneorganicheskiye soyedineniyya (New Oxygen-Containing Silicon Organic Compounds) Baku, Azerneftneshr, 1960. 190 p. Errata slip inserted. 1,000 copies printed.
- Ed. (Title page): Yu. G. Mamedaliyev, Academician of the Academy of Sciences Azerbaydzhanskaya SSR, Professor; Ed. of Publishing House: A. S. Shteyngel'.
- PURPOSE: This book is intended for persons working with organosilicon compounds, and for students in schools of higher education.
- COVERAGE: The book deals with the chemistry of organosilicon compounds, including their synthesis and conversion. It describes the use of organosilicon compounds in the manufacture of heatresistant and electric insulating materials, anticorrosive

Card-1/7

New Oxygen-Containing Silicon (Cont.) SOV/4989

coatings, silicon rubber, lubricants, and hydrophobic and gluing materials. The book also describes the new oxygen-containing organosilicon compounds such as organosilicon alcohols and silanols, simple vinyl ethers and their derivatives, and organosilicon acetals and acylals. The production and properties of chlorosilanes are described briefly. B. N. Dolgov, K. A. Andrianov, and A. P. Kreshkov are cited as Soviet writers on the chemistry of organosilicon compounds. References accompany each chapter.

TABLE OF CONTENTS:

Introduction	3
Ch. I. Alkyl (Aryl) Chlorosilanes 1. Production of alkyl (aryl) chlorosilanes A. Organometallic synthesis of saturated alkyl (aryl) chlorosilanes containing chlorine directly at the silicon atom	4 4

Card 2/7

SHOSTAKOVSKIY, M.F.; SHIKHIYEV, I.A.; KOMAROV, N.V.

A.E. Favorskii; on the centennial of his birth. Azerb.khim. zhur. no.2:69-73 '60. (MIRA 14:8) (Favorskii, Aleksei Efgrafovich, 1860-1945)

S/079/60/030/009/005/015 B001/B064

5.3700 AUTHORS: 2209.

Shikhiyev, I. A., Aliyev, M. I., Aslanov, I. A.,

Karayeva, Sh. V.

TITLE: Investigations in the Field of the Synthesis and Conversion

of Unsaturated Organosilicon Compounds. VII. Synthesis and Properties of Some Secondary and Tertiary y-Silicon-con-

taining Acetylene Alcohols

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol. 30, No. 9,

pp. 2916-2919

TEXT: In the present paper (Ref. 1), the authors describe a method of synthesizing mono-, bi-, and trivalent γ -silicon-containing tertiary alcohols of the acetylene series. The present investigation deals with the synthesis of some representatives of the γ -silicon-containing secondary and tertiary acetylene alcohols by allowing the respective organomagnesium compound of the acetylene series to react with trialkyl chlorosilanes. The presence of the hydroxyl group in the γ -silicon-containing acetylene alcohols was confirmed by acetylation (Ref. 2) by the scheme given (details

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Investigations in the Field of the Synthesis and Conversion of Unsaturated Organosilicon Compounds. VII. Synthesis and Properties of Some Secondary and Tertiary \gamma-Silicon-containing Acetylene Alcohols

S/079/60/030/009/005/015 B001/B064

in the experimental part). The following γ -silicon-containing acetylene alcohols were described: 1-trimethyl silyl-3-methyl pentin-1-ol-3;, 1-dimethyl ethyl silyl-3-methyl pentin-1-ol-3; 1-trimethyl silyl pentin-1-ol-3; 1-trimethyl silyl-3-methyl hexine-1-methyl-5-ol-3; 1-trimethyl silyl-3-methyl heptin-1-ol-3; 1-tri-methyl silyl hexin-1-ol-3. The presence of a hydroxyl group in the alcohols obtained was confirmed by the following silicon-containing acetals synthesized from them: n-butyl trimethyl silyl methyl pentine-, n-butyl dimethyl ethyl silyl methyl pentine-, n-butyl triethyl silyl methyl pentine-, n-butyl trimethyl silyl methyl hexine-, n-butyl trimethyl silyl methyl hexine-, n-butyl trimethyl silyl methyl heptine-, and n-butyl trimethyl silyl hexine acetal. The alcohols and acetals obtained are given together with their constants in a table. There are 1 table and 2 Soviet references.

ASSOCIATION:

Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Petroleum-chemical Processes of the Academy of Sciences Azerbaydzhanskaya SSR)

Card 2/3

Investigations in the Field of the Synthesis and Conversion of Unsaturated Organosilicon Compounds. VII. Synthesis and Properties of Some Secondary and Tertiary \u03c4-Silicon-containing Acetylene Alcohols

\$\dot{079/60/030/009/005/015} B001/B064

SUBMITTED:

September 21, 1959

Card 3/3

s/081/62/000/016/011/043 B168/B186

Shikhiyev, I. A., Aliyev, M. I., Guseyn-Zade, B. Kh.,

AUTHORS:

Karayeva, Sh. V. Synthesis of acetylene alcohols containing V-silicohydride

and their dehydrocondensation by dimethylphenylsilanol TITLE:

Referativnyy zhurnal. Khimiya, no. 16, 1962, 235, abstract 16Zh271 (Azerb. khim. zh., no. 3, 1961, 67-70 [summary in PERIODICAL:

TEXT: Production of RR'C(OH)C = CSiHR2" (I, where R = CH3, C2H5;

 $R' = CH_3$, C_2H_5 , tert- C_4H_9 ; $R'' = CH_3$, C_2H_5) by the reaction of

RR'C(OMgBr)C = CMgBr with R2"SiHCl (II) is described. The reaction of I

with $C_6H_5(CH_3)_2SiOH$ (III) produces $RR^{\dagger}C(OH)C = CSi(R_2")OSi(CH_3)_2C_6H_5$ (IV) with liberation of H2. The presence of an OH group in I is proved by

acetalization and by the fact that the corresponding siloxy derivatives are

Card 1/3

Synthesis of acetylene alcohols...

S/081/62/000/016/011/043 B168/B186

produced in accordance with the formula: I + CH₂ = CHOC₄H₉ (V) — CH₃CH(OC₄H₉)OC(RR')C = CSiHR₂" (VI). 0.2 mole II (R₂" = CH₃ and C₂H₅) is gradually added, during cooling, to Iotsich's reagent (consisting of 0.4 mole C₂H₅Br, 0.4 mole Mg and 0.2 mole methyl-tert-butylacetylenyl-carbinol); after 12 hr this mixture is heated for 6 hr, after 4 hr (20°C) it is decomposed with dilute HCl and I (R = CH₃, R' = tert-C₄H₉, R₂" = CH₃ and C₂H₅) (Ia) (here and henceforth yield in %, boiling point in °C/mm, n^{2O}D, d₄ owill be given for isolated substances), 26.3, 69/2, 1.4603, 0.8768, is isolated from the ester layer. 0.01 g ZnCl₂ is added to a mixture of 0.05 mole Ia and 0.05 mole III in C₆H₆; when evolution of H₂ has ceased the C₆H₆ is driven off and IV (R = CH₃, R' = tert-C₄H₉, R₂" = CH₃ and C₂H₅), 21.55, 106/0.18, 1.5124, 0.9842, is isolated from the residue. 0.2 ml 33 % HCl is added to a mixture of 0.03 mole I (R = R' = CH₃, R₂" = CH₃ and C₂H₅) and 0.03 mole V; this is heated for Card 2/3

Synthesis of acetylene alcohols...

S/081/62/000/016/011/043 B168/B186

30 min at 70° C and neutralized after 12 hr with calcined K_2 CO₃, and VI (R = R' = CH₃, R₂" = CH₃, C₂H₅), 26.04, 119/4, 1.4422, 0.8725, is isolated from it. Other representatives of this class of compound are produced in a similar manner. [Abstracter's note: Complete translation.]

Card 3/3

s/081/62/000/018/056/059 B168/B186 Shikhiyov, I. A., Aliyev, M. I., Sadykhzade, S. I., Shchegol', Sh. S., Patliyev, S. B., Akhundova, G. Yu., Krasnokutskiy, V. P., Guseynova, M. A., Mukharamova, Kh. F., Murbanaliyeva, AUTLORS: T. Kh., Nikolayeva, L. Synthesis and use of silico derivatives of naphthenic acids in the production of divinylstyrene rubber Referativnyy zhurnal. Khimiya, no. 18, 1962, 559, abstract TITLE: 19P485 (Amerb. khim. zh., no. 5, 1961, 65 - 68 [Summary in PERIODICAL: TEXT: Dimethyldichlorosilane reacting with naphthenic acids in the presence of trieth, limine produces dimethyldicarboxy-bis-cycloalkylsilane The product was used as filler for butadienestyrene The Defoe number of the raw outa-.(CH₃)251(OCOC₁₀H₂₀)2. dienestyrene rubber was 2000, thermal plastification time 40 min, Defoe rubber instead of lubricating oil 18. number of the plasticized butadienestyrene rubber 1100, tensile strength card 1/2

Synthesis and use of ...

S/081/62/000/013/056/059
B168/B186

250 kg/cm², elasticity 31%, relative elongation 522% and permanent elongation 16%. [Abstracter's note: Complete translation.]

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Card 2/2

S/079/61/031/001/004/025 B001/B066

5 3700

2200

AUTHORS:

Shikhiyev, I. A., Aliyev, M. I., Aslanov, I. A., and

Garayeva, Sh. V.

TITLE:

Studies in the Field of the Synthesis and Conversions of Unsaturated Organosilicon Compounds. VIII. Synthesis and Properties of Some Ditertiary y-Silicon-containing Acetylene

Glycols

PERIODICAL: Zhurnal obshchey khimii, 1961, Vol. 31, No. 1, pp. 35 - 38

TEXT: In Refs. 1 and 2, the authors studied the reaction of dimagnesium bromo dimethyl-ethinyl carbinol with dialkyl-(aryl)-dichloro silanes in the presence of catalytic amounts of copper- and mercury chlorides according to the equation

$$\begin{array}{c} \text{CH}_{3} \\ \text{2BrMgO} - \overset{\text{CH}}{\text{C}} & \text{C} \equiv \text{CMgBr} + \text{R}_{2}\text{SiCl}_{2} \longrightarrow \text{R}_{2}\text{Si} \\ \text{CH}_{3} \\ \end{array}$$

Card 1/3

Studies in the Field of the Synthesis and Conversions of Unsaturated. Organosilicon Compounds. VIII. Synthesis and Properties of Some Ditertiary Y-Silicon-containing Acetylene Glycols

S/079/61/031/001/004/025 B001/B066

The presence of two hydroxyl groups in the resultant ditertiary, V-siliconcontaining acetylene glycols was confirmed by their conversion to the
corresponding acetyl derivatives. The present paper describes the synthesis of some other branched ditertiary V-silicon-containing acetylene
glycols, the structure of which was also confirmed by conversion to the
corresponding acetyl derivatives (Ref. 3) (Table). The following six new
compounds of ditertiary, V-silicon-containing acetylene glycols were
synthesized: bis-(3-methyl-pentin-1-ol-3)-dimethyl silane, bis-(3-methylpentin-1-ol-3)-methyl-ethyl silane, bis-(3-methyl-pentin-1-ol-3)-methylpropyl silane, bis-(3,5-dimethyl-hexin-1-ol-3)-dimethyl silane, bis(3-methyl-heptin-1-ol-3)-dimethyl silane, bis-(3-methyl-heptin-1-ol-3)diethyl silane. The presence of two hydroxyl groups in ditertiary,
V-silicon-containing acetylene glycols was confirmed by the following
new acetyl derivatives obtained from them; bis-(3-methyl-propyl-1-acetoxy-3)-dimethyl silane, bis-(3-methyl-pentine-1-acetoxy-3)-methyl-propyl
silane, bis-(3,5-dimethyl-hexine-1-acetoxy-3)-dimethyl silane, bis-

Card 2/3

Studies in the Field of the Synthesis and Conversions of Unsaturated Organosilicon Compounds. VIII. Synthesis and Properties of Some Ditertiary y-Silicon-containing Acetylene Glycols

S/079/61/031/001/004/025 B001/B066

(3-methyl-heptine-1-acetoxy-3)-dimethyl silane, and bis-(3-methyl-heptine-1-acetoxy-3)-diethyl silane. There are 1 table and 3 Soviet references.

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk Azer-

baydzhanskoy SSR (Institute of Petrochemical Processes of the

Academy of Sciences Azerbaydzhanskaya SSR)

SUBMITTED:

February 15, 1960

Card 3/3

S/079/61/031/011/008/015 D228/D305

5.3700

AUTHORS:

Shikhiyev, I. A., Aslanov, I. A., and Yusufov, B. G.

TITLE:

Synthesis and conversion of primary and secondary

monoatomic acetyl- 7 -germanium alcohols

PERIODICAL:

Zhurnal obshchey khimii, v. 31, no. 11, 1961, 3647~3648

TEXT: The authors give the first details of the synthesis and certain properties: 1-triethylgermanopropin-1-o1-3 — Et₃GeC&CMeO (I); 1-triethylgermanohexin-1-o1-3 — Et₃GeC&CCHOHPr (II); n-butyltriethyl-propylgermanium acetal — MeCH(0Bu0CH₂)C&CGeEt₃ (III); and n-butyltriethylhexylgermanium acetal — MeCH(0Bu)0(Pr)CHG&CGeEt₃ (IV). In previous research, I. A. Shikhiyev, M. F. Shostakovskiy, I. A. Aslanov, and N. V. Komapov (Ref. 1: Zh. obshch. khimii, 29, 1549, 1959; Ref. 2: Usp. khims, 27, 1504, 1958) developed a method for preparing monomy duomand triatomic acetyl-y-germanium alcohols and proved the presence of

Card 1/2

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30187

Synthesis and conversion...

S/079/61/031/011/008/015 D228/D305

hydroxyl groups in these compounds by their conversion to the corresponding acetals. The synthesis of I and II entails the reaction of propinol with the Grignard reagent; the mixing of the solution for 3 hr. as it is cooled to ~5°; the addition of triethylgermanium chlorides the solution of the residue after about 8 ~ 12 hr. in water and dil. IICls the removal of the ether layer; and the double multiple distillation of the remaining solution when the desired alcohols boil over at 107. 108° and 110 ~ 111° respectively. III and IV are prepared by stirring a mixture of I and II with vinylbutyl ether and HCl which is then heated for 1 hr. at 35 - 95° and allowed to stand overnight. After neutralization and removal of the ether, the acetals are obtained by double multiple distillation, their respective boiling-points being 146 - 148° and 152°. There are 1 table and 2 Soviet-bloc references.

SUBMITTED:

December 27 1960

Card 2/2

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•			D228/D305	
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	53700	Shikhiyev, I. A. M. Guseynzade, B. M.	D223/D305 A. I., Garayava, Sn. V., and Silicoorganic acetyl sicohol	
	AUTHOR	useis of branches	1961 3649-36	52
	TITLE:	and Stanchey khimily	s the synthesis (1)	\wedge
	PERIODICAL;	and glyton Zhurnal obshehey khimii authors give the first desc ilyl-3-ethylpentyn-4-ol-3 ilyl-2,2,3-trimethylpentyn-4 ilyl-2,2,3-trimethylpentyne acetal ethylsilylethylpentyne acetal	MeCH ₂ C(Et)OHC (CSIMe3	(11)3
	5-trimethyls	ily1-2,2,3-trimethy1pentyne acetalethy1sily1ethy1pentyne acetalethy1sily1trimethy1pentyne thy1pentyne thy1sily1trimethy1penty1trimethy1sily1trimethy1sily1trimethy1sily1trimethy1sily1trimethy1sily1trimethy1sily1trimethy1pentyne thy1sily1trimethy1pentyne thy1sily1trimethy1pentyne acetalethy1sily1trimethy1pentyne acetalethy1sily1trimethy1pentyne acetalethy1sily1trimethy1pentyne acetalethy1sily1trimethy1pentyne acetalethy1sily1trimethy1	ntyne acetal ntyne acetal is (3 ethylpentyn-leoleol	silyh
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Synthesis of branched...

S/079/61/031/011/009/015 D228/D305

2.2.3 trime thylpen tyn-4-01-3) dime thyl silane $-\int M_{3}CC(M_{2})OHCC_{2}$ siMe₉ (VI); and bis (3 thy) pentyon cacetoxy-3) dimethylsilane Mechac(Et)(Ocome)Coc_2sime_2 (VII) Their work is a continuation of previous research by I. A. Shikhiyer, M. F. Shostakovskiy, N. V. Komarow, M. I. Aliyev, I. A. Aslanov and Sh. V. Garayeva (Ref. 1: Nowyye kislorodo soderzhashchiye kremneorganicheskiye soyedineniya (New Oxygen-Containing Silicoorganic Compounds). Baku, 1960; Ref. 2: Zh. obshch. khimii, 30; 2916, 1960), in which it was shown that silicoorganic acetyl alcohols and glytols are formed through the reaction of trialkyl(azyl)chlorosilanes with dimethylacetylcarbinyldimagnesium bromid; in the presence of a CuCl and HgCl, catalyst, I. A. Shikhiyev, N. V. Komarow and I. A. Aslanow (Ref. 4: Usp. Khim., 27_{\odot} 1504, 1958) also established the structure ture of these compounds by hydrogenation and acetalization. The method of T. A. Favorskaya and I. A. Favorskaya (Ref. 5. Zh. obsheb. khimii, 10. 451, 1940) was used to prepare I. This entails the stirring and cooling of a solution of the Grignard reagent and disthylacetylcarbinol for 3 hr; the addition of trimethylchlorosilans, followed by the beating of Card 2/3

Synthesis of branchedo....

S/079/61/031/011/000/015 D228/D305

the solution and its treatment with dil. HCl; the separation of the ether and water layers; and distilling-off the required alcohol at 69 - 70° o V was obtained by gradually adding dimethyldichlorosilane to a solution of the Grignard reagent and diethylacetylcarbinol, which was first cooled and stirred for 2 hr. The solution was allowed to stand overnight, after which dil. HCl was added, and the ether and water layers were then separated; the desired compound boils over at 128 130° during double multiple distillation. II and VI were synthesized by the same procedure adopted for I and V. The authors consider the presence of hydroxl groups in alcohols I and II and glycol V to be proved by the respective conversion of these compounds into acetals III and IV and acytal VII. In the case of III (bopo 95 - 96°) and IV (bopo 95 - 91°), the conversion was effected with vinylbutyl ether and HCl, while VII (bopo 148 - 149°) was obtained from V by means of acetic anhydrides. There are I table and 5 Soviet bloc references.

Card 3/3

5.3700

27262 s/020/61/139/005/015/02 B103/B217

AUTHORS:

Shikhiyev, I. A., Aliyev, M. I., and Guseynzade, B. M.

TITLE:

Studies of synthesis and conversions of unsaturated organosilicon compounds. Synthesis of branched silicon

hydrocarbons of the diacetylene series

PERIODICAL:

Akademiya nauk SSSR. Doklady, v. 139, no. 5, 1961,

1138-1140

TEXT: The authors continue their studies in the field mentioned in the title, and describe a new method of synthesizing the initially mentioned compounds on the basis of halogen derivatives of organosilicon acetylene alcohols and of tert-butyl acetylene according to the enclosed scheme,

 $R-COH-C \equiv C-Si(CH_0)_2 \xrightarrow{HCI} R-CCI-C \equiv C-Si(CH_0)_2 \xrightarrow{(CH_1)_1C-C \equiv CMgBr}$ CH_3 CH_3 CH_4 $CH_3)_3C-C \equiv C-C-C \equiv C-Si(CH_2)_3,$

Card 1/4

"APPROVED FOR RELEASE: 08/25/2000

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27262 S/020/61/139/005/015/021 B103/B217

Studies of synthesis and conversions ...

where $R = CH_3$; C_2H_5 ; and $tert-C_4H_9$. The existence of two triple bonds in the diacetylene silicon hydrocarbons obtained was proved by hydrogenation of the latter up to saturation. The authors synthesized and characterized for the first time: the three representatives of branched diacetylene silicon hydrocarbons whose constants are given in Table 1. A three-necked flask with reflux cooler served for synthesis. 5-trimethyl silyl-3-methylpentyn-4-ol-3 through which gaseous HCl was bubbled was used for the synthesis of 5-trimethyl dilyl-3-methylpentyne-4-chlorine-3. Synthesis of 1-trimethyl $silyl-3,6\beta$ -trimethyl-3-ethyl heptadiine-1,4: Tert-butyl acetylene (8.5 g) in 20 ml ether was added to a Grignard reagent during 25 min under continuous stirring and cooling. After 40 hr standing, the content of the flask was heated on a water bath up to 35°C, and kept at this temperature until the ethane separation ceased. 1 g $\mathrm{Cu_2Cl_2}$ and 0.5 g $\mathrm{HgCl_2}$ were added as catalyst to the resulting complex magnesium bromine tert-butyl acetylene. After 0.5 hr stirring, the mixture was cooled down to -2°C, and 21.68 g acetylene chloride added. Then, the mixture was stirred during 58 hr at room temperature, heated during 6 hr, and decomposed by diluted HCl. An ether solution and the extract were dried over calcined Na2SO1. After Card 2/4

27262 s/020/61/139/005/015/021 B103/B217

Studies of synthesis and conversions

distilling off the ether, 14 g of the final product was isolated by double distillation. Two further representatives of the said compounds were obtained in similar manner. The authors thank I. F. Zhukova for assisting in the hydrogenation of diacetylene silicon hydrocarbon at the laboratory of Professor L. Kh. Freydlin. Raney-Ni (0.2 g) in 5 ml methanol was used. The mixture was saturated with hydrogen, and then 0.123 silicon hydrocarbon introduced. Altogether 43.6° ml hydrogen was absorbed. 47 ml hydrogen is theoretically necessary for complete hydrogenation of two triple bonds. There are 1 table and 3 Soviet-bloc references.

ASSOCIATION:

Institut neftekhimicheskikh protsessov Akademii nauk AzerbSSR (Institute of Petrochemical Processes of the Academy of Sciences of the Azerbaydzhanskaya SSR)

PRESENTED:

January 16, 1961 by B. A. Arbuzov, Academician

SUBMITTED:

January 11, 1961

Card 3/4

s/079/62/032/005/006/009 D204/D307

Shikhiyev, I.A., Aliyev, M.I., and Guseynzade, B.Kh. Studies of the syntheses and transformation of oxygencontaining organosilicon to compounds. XI. Synthesis of AUTHORS:

symmetrical organosilicon trichloroacetals TITLE:

Zhurnal obshchey khimii, v. 32, no. 5, 1962, 1646-1647 TEXT: CCl₃CH(OSiMe₃)₂ (I) was prepared in 18.1 % yield from chloral hydrate (0.22 moles) and MegSiCl (0.4 moles), in ether/EtgN, at room temperature, over 1 hr. CCl₃CH / OSiEt₃ / 2 (II) was synthesi-PERIODICAL:

zed by an analogous reaction between chloral hydrate and EtaSiCl; Et_SiOSiEt_3 and Et_3SiOCHOCHOSiEt_3 (III) were also present in the

reaction mixture. The structure of II was demonstrated by synthesizing it. in 17.4 % vield from chloral hydrate (0.1 mole) which reaction mixture. The structure of il was demonstrated by synthes, which zing it, in 17.4 % yield, from chloral hydrate (0.1 mole), which had been refluxed with dry henzene generating the water formed. Nad been refluxed with dry benzene separating the water formed, and Card 1/2

Card 1/2

Studies of the syntheses and ...

S/079,62/032/005/006/009 D204/D307

 $\mathrm{Et_{3}SiOH}$ (0.03 moles), in presence of HCl (33 %, 2 drops), on heating to 80 - 82°C. The mixture was then allowed to stand overnight, was treated with 1 drop of HCl, heated for a further 2 hrs., neutralized with KOH and distilled under vacuum. A number of acetylenic organosilicon acetals was prepared by the latter method, whose projecties shall be described in future publications. Compound II decomposed into Et SiOSiEt and III on repeated distillation. Compounds I, II and III are new.

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Petro-Chemical

Processes of the Academy of Sciences of the Azerbayd-

zhan SSR)

SUBMITTED: May 3, 1961

Card 2/2

SHIKHIYEV, I.A.; ALIYEV, M,I.; ISRAFILOVA, S.Z.

Synthesis and conversions of organophosphorus compounds. Part 2: Synthesis of some derivatives of methylcyclohexylphosphinic acid. Zhur.ob.khim. 32 no.8:2686-2688 Ag '62. (MIRA 15:9)

1. Institut neftekhimicheskikh protsessov AN Azerbaydzhanskoy SSR. (Phosphinic acid)

SHIKHIYEV, I.A.; ASLANOV, I.A.; YUSUFOV, B.G.

Synthesis and conversions of oxygen-containing unsaturated organogermanium compounds. Part 15: Synthesis of mono-and diatomic tertiary γ -germanium acetylenic alcohols and some of their derivatives. Zhur.ob.khim. 32 no.10:3148-3151 0 '62. (MIRA 15:11)

(Alcohols)
(Germanium organic compounds)

43310

S/079/62/032/011/005/012 D204/D307

5.3700

Shikhiyev, I.A., Guseynzade, B.M., and Aliyev, M.I.

TITLE:

AUTHORS:

Investigations of the synthesis and transformations of oxygen-containing organic and organosilicon compounds. XIV. Organic and organosilicon derivatives of chloral

hydrate

PERIODICAL: Zhurnal obshchey khimii, v. 32, no. 11, 1962,

3630 - 3633

TEXT: The present study was aimed at the consideration of the interactions of chloral hydrate (A) with organic and organosilicon tertiary acetylenic alcohols and simple alkyl vinyl ethers. The products consisted of the corresponding symmetrical trichloroacetals or unsaturated simple ethers of A, depending on the conditions and structures of the reagents concerned. Thus ethoxyethyl-hydroxytrichloroethyl ether of 2,2,2-trichloroethanediol-1,1 (I) was prepared by treating a solution of A in benzene with ethyl vinyl ether cooling, adding a small drop of 33 % HCl, heating to 60-70°C for 1 hr., leaving overnight, neutralization and distillation. Ethoxy-Card 1/2

S/079/62/032/011/005/012 D204/D307

Investigations of the synthesis ...

ethyl-triethylsiloxytrichloroethyl ether of 2,2,2-trichloroethane-diol-1,1 (II) was synthesized (to show the presence of an OH group in I), by treating a solution of I in pyridine with triethylchlorosilane, heating to 50 - 60°C for 1 hr., filtering and vacuum distilling the filtrate. Trimethylsilyl-(dimethylpropynyl) ether of 2,2,2-trichloroethanediol-1,1 (V) was analogously prepared from hydroxydimethylpropynyl ether of 2,2,2-trichloroethanediol-1,1 (IV) with trimethylchlorosilane. Bis-butoxyethyl ether of 2,2,2-trichloroethanediol-1,1 (III) was obtained by heating a benzene solution of A with butyl vinyl ether, cooling, adding 1 drop of 33 % HCl, heating to 60-70°C for 1 hr., neutralizing and distilling in vacuum. Compound IV and also butoxyethyl-dimethylpropynyl, - bis(trimethylsilyl) dimethylpropynyl, - hydroxy-methyl test. - butyltrimethyl-silylpropynyl, - and bis-dimethylvinylpropynyl - ethers of 2,2,2-trichloroethanediol-1,1 were prepared by analogous reactions. There is 1 table.

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk

Azerbaydzhanskoy SSR (Institute of Petrochemical Pro-

cesses of the Academy of Sciences Azerb. SSR)

SUBMITTED: December 11, 1961

Card 2/2

SHIKHIYEV, I.A.; ALIYEV, M.I.; MUKHARAMOVA, Kh.F.; ISRAYELYAN, D.R.

Synthesis of alkyl chlorosilanes based on the pentaneamylene fraction. Azerb. khim. zhur. no.2:21-24 '63. (MIRA 16:8)

ACCESSION NR: AP4022010

S/0249/63/019/012/0015/0017

AUTHOR: Shikhiyev, I. A.; Aliyev, M. I.; Zeynalov, B. K.; Israyelyan, D. R.; Mukharamova, Kh. F.

TITLE: Synthesis of vinyl esters from commercial fractions of C_5 - C_6 fatty acids and acetylene

SOURCE: AN AzerbSSR. Doklady*, v. 19, no. 12, 1963, 15-17

TOPIC TAGS: vinyl ester, C5 fatty acid, C6 fatty acid, activated carbon

ABSTRACT: The purpose of the present investigation was to find an industrial application for the fatty acids of the C₅ - C₆ fraction obtained by direct oxidation of the paraffin hydrocarbons. A commercial fraction of fatty acids (boiling point of 180—200C) was reacted with acetylene, using activated AG carbon impregnated with salts of the same acids as catalyst. The equipment consisted of a reactor, a coil vaporizer, and five condensers, two of which were cooled with dry ice. The experiments were carried out at a acetylene:acid ratio of 9:1. The temperature of the catalyst was 245—250C, the temperature in the coil vaporizer for fatty acids was 220—225C, the rate

ACCESSION NR: AP4022010

of fatty acid feed was 30 cm³/hour, and the acetylene rate was 3.3—3.5 liter/min. The acetylene was mixed with the vapors of fatty acids before entering the reaction chamber. An 81% yield of the catalyzate was obtained, with the noncondensed vapors being discharged into the atmosphere. Five fractions were isolated from the catalyzate within a boiling range of 85—180C (75.9% were vinyl esters, the bromine number of which ranged from 125.7 to 84.44). The 135—155C fraction was the largest, representing a 31.3% yield on the basis of the fatty acids used in the reaction. It had a molecular weight of 131.4 and a bromine number of 112.3, as against a theoretical bromine number of 118.4 for vinyl ester. Orig. art. has: 1 table.

ASSOCIATION:

Im. Yu. G. Mamedaliyeva INKhP

SUBMITTED: 120ct63

ATD PRESS: 3045

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SUB CODE: OC

NO REF SOV: 010

OTHER: 000

Card 2/2

S/079/63/033/002/003/009 D204/D307 AUTHORS: Shikhiyev, I.A. and Aslanov, I.A. TITLE: Studies of the synthesis and conversions of unsaturated organosilicon compounds. XVI. Synthesis and properties of some secondary Y-silicon-containing acetylenic alcohols PERIODICAL: Zhurnal obshchey khimii, v. 33, no. 2, 1963, 377 - 379 A continuation of earlier work (Novyye kislored-TEXT: soderzhashchiye kremneorganicheskiye soyedineniya New oxygen-containing organosilicon compounds _7, Baku (1960)). 1-Dimethylethylsilylhexyn-1-01-3 (I) was prepared by reacting EtMgBr with propylethynyl-carbinol in the cold (~00C), stirring for 5 hrs at room temperature, and treating with EtMe₂SiCl. The complex was then decomposed with HCl. Compounds R₃SiC = C-CHOH-C₃H₇ (where R₃ = CH₃, (C₂H₅)₂ (II) or (C₂H₅)₃ (III) and R_2 SiHC = C-CHOH- C_3H_7 (where R_2 = (CH_3)₂ (IV); R_2 = CH_3 , Card 1/2

Studies of the synthesis S/079/63/033/002/003/009 C2H5 (V); R2 = (C2H5)2 (VI); R2 = CH3, C3H7 (VII); R2 = (1so-C3H7)2 (VIII); R2 = (C3H7)2 (IX)) were obtained by an analogous procedure, using R2SIHC1 for compounds IV = IX. To demonstrate the presence of the OH group, n-butyldimethylethylsilylhexyne acetal (X) was prepared from I and vinylbutyl alcohol in the presence of some 33 % HC1. The temperature was allowed to rise to 550C and was then held at 90°C for 30 min. Similar acetals were prepared from the other alcohols. All these compounds are new. There is 1 table. SUBMITTED: November 5, 1961				
The state of the old group, n-butyldimethylatilylhexyne acetal (X) was prepared from I and vinylbutyl alcohol in the presence of some 35 % HCl. The temperature was allowed to rise to 550C and was then held at 90°C for 30 min. Similar acetals were prepared from the other alcohols. All these compounds are new. There is 1 table. SUBMITTED: November 5, 1961	Studies of the syn	S/079/63/0 hthesis D204/D307	33/002/003/009	
SUBMITTED: November 5, 1961	procedure, using R presence of the OH was prepared from 33 % HCl. The tempat 90°C for 30 min	= (C ₃ H ₇) ₂ (IX)) were obtained 2SIHCl for compounds IV - IX. group, <u>n</u> -butyldimethylethyls I and vinylbutyl alcohol in the erature was allowed to rise to . Similar acetals were prepare	by an analogous To demonstrate the ilylhexyne acetal e presence of some 55°C and was then d from the other a	(x)
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SHIKHIYEV, I.A.; VATANKHA, A.A.; RZAYEVA, S.A.; GUSEYNZADE, B.M.

Synthesis and transformations of oxygen-containing organic and organosilicon compounds. Azerb. khim. zhur. no.5:27-30 '64. (MIRA 18:3)

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Pc-li/Pr-li RM S/0316/64/000/006/0033/0037

28 B

ACCESSION NR:
AUTHOR: Shikh

Shikhiyev, I. A.; Verdiyeva, S. Sh.; Aslanov, I. A.

TITLE: Studies in the field of the synthesis and conversions of oxygen-containing organic and organosilicon compounds. Synthesis of organic and organosilicon trichloroacetals and incomplete acetylenic ethers of chloral hydrate

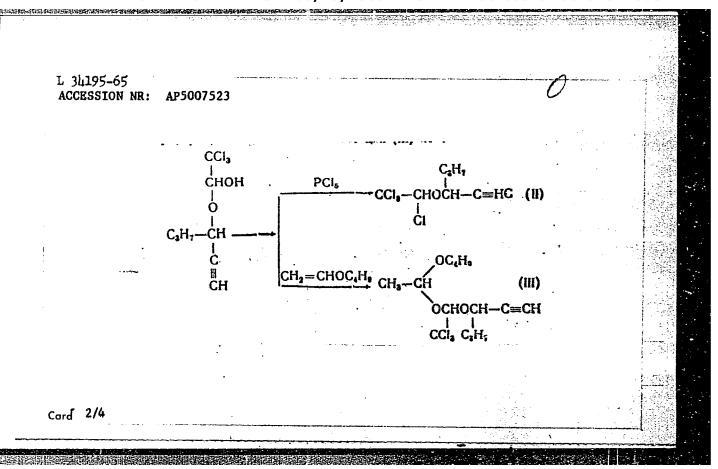
SOURCE: Azerbaydzhanskiy khimicheskiy zhurnal, no. 6, 1964, 33-37

TOPIC TAGS: silicoorganic compound, chloral hydrate ether, acetylenic ether, trichloroacetal, organosilicon trichloroacetal

ABSTRACT: The incomplete ether of chloral hydrate (I) and an acetylenic &-chloro ether (II) were obtained and characterized for the first time, and the presence of a hydroxyl group in (I) was established by the formation of the corresponding acetal (III):

CCI₃-CH
$$C_3H_1$$
 C_3H_7 C_3H_7

Card 1/4



L 34195-65

ACCESSION NR: AP5007523

The presence of a triple bond in acetal (III) was established by the formation of the corresponding organosilicon acetal (IV).

$$CH_{3}-CH$$

$$CH_{3}-CH$$

$$CCI_{3}$$

$$CCI_{3}$$

$$CH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$CCH_{3}-CH$$

$$(IV)$$

The experimental procedures employed are fully described, and the physicochemical constants of the products obtained are tabulated. Orig. art. has: 1 table and 12 formulas.

Card 3/4

L 3\(\frac{1}{95}-65\)
ACCESSION NR: AP5007523

ASSOCIATION: None

SURMITTED: 00 ENCL: 00 SUB CODE: 0C

NO REF SOV: 004 OTHER: 000

ENG(j)/ENT(m)/EPF(o)/ENP(j)/T/EWP(t)/ENP(b)/ENA(h)/ENA(1) Pr-4/Peb IJP(c) JD/RM ACCESSION NR: AP5007719 3/0249/64/020/011/0013/001 AUTHORS: Shikhiyev, I. A.; Guseynzade, B. M.; Abdullayev, H. D. TITLE: Investigations of gamma synthesis and conversion of unsaturated oxygenbearing silicon and germanium organic compounds SOURCE: AN AzerbSSR Doklady, v. 20, no. 11, 1964, 13-17 TOPIC TAGS: silicon organic polymer, germanium compound, organic derivative, acetylene alcohol ABSTRACT: This is a continuation of the authors work in the field of heteroorganic derivatives of acetylene. The present paper concerns the gamma synthesis and conversion of unsaturated oxygen-bearing silicon- and germanium-organic compounds. A method has been developed for producing silicon- and germanium-organio monatomic diacetylene alcohols from diatomic silicon-organic alcohol, and also by reaction between the Iotsich group of some tertiary acetylene alcohols with Ysilicon- and germanium-organic chlorides. The presence of the hydroxyl group in silicon-organic monatomic diacetylene alcohols is demonstrated by dehydration and by cyanethylation. The investigations yielded six different representatives of silicon- and germanium-organic monatomic discetylene alcohols and their derivatives. These are described for the first time, and their constants are tabulated in the Card 1/2

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Card 2/2					

ACCESSION NR: AP4018053

s/0079/64/034/002/0394/0396

AUTHOR: Shikhiyev, I. A.; Guseynzade, B. M.; Mekhmandarova, N. T.; Aslanov, I. A.

TITLE: Research in the area of synthesis and conversion of unsaturated silicon germanium organic compounds 17. Synthesis and some conversions of silicon and germanium organic alcohols of the diacetylene series

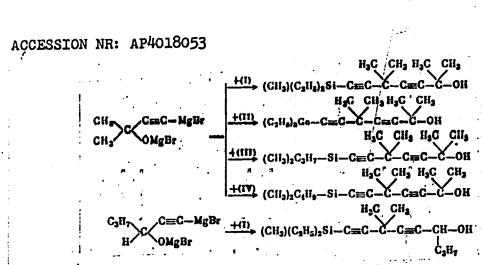
SOURCE: Zhurnal obshchey khimii, v. 34, no. 2, 1964, 394-396

TOPIC TAGS: silicon germanium, synthesis unsaturated silicon germanium, conversion unsaturated silicon germanium, organic alcohol, diacetylene series organic alcohol

ABSTRACT: The synthesis of silicon and germanium organic acetylene chlorides is studied by means of a reaction of gaseous hydrogen chloride with corresponding acetylene alcohols according to:

Card 1/4

ACCESSION NR:	AP4018053					······································		•
(CI	H ₃) ₂ COH—C=C—Ge((CH ₃) ₂ COH—C=C R ₂ Si=Si(CH ₂)(C ₁ E	-SIR	CH ₃) ₂ CCi	-C=C-SIR				
Silicon and g	ermanium org	ganic mone	patomic	diacetyl	ene alcoho	ols with	h S the	
isolated trip corresponding and germanium	: Iotsich rea	igent of a	acetylen	e alcoho	ls with so	ome sil:	icon	3
corresponding	: Iotsich rea	igent of a	acetylen	e alcoho	ls with so	ome sil:	icon	
corresponding	: Iotsich rea	igent of a	acetylen	e alcoho	ls with so	ome sili	icon	



Four representative silicon and germanium organic acetylene tertiary chlorides are described for the first time: 4-methyldiethylsilicon-2-chlor-2-methylbutine-3; 4-triethylgermanium-2-chlor-2-methylbutine-3; 4-dimethylpropylsilicon-2-chlor-2-methylbutine-3; 4-dimethylbutyl-silicon-2-chlor-2-methylbutine-3. Five representative silicon and

Card 3/4

ACCESSION NR: AP4018053

germanium organic monoatomic diacetylene alcohols determined for the first time are also described: 9-methyldiacetylsilicon-7,7-dimethyl-nonadiine-5, 8-ol-4; 7-methyldiethylsilicon-2,5,5-trimethylheptadiine-3,6-ol-2; 7-triethylgermanium-2,5,5-trimethylheptadiine-3,6-ol-2; 7-dimethylpropylsilicon-2,5,5-trimethylheptadiine-3,6-ol-2; 7-dimethylbutylsilicon-2,5,5-trimethylheptadiine-3,6-ol-2. The germanium organic diacetylene acetal n.-butyltriethylgermaniumtetramethylhexadiine-acetal is described for the first time. Orig. art. has: 2 tables.

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Petrochemical Processes, Academy of Sciences Azerbaizan SSR)

SUBMITTED: 19Dec62

DATE ACQ: 19Mar64

ENCL: 00

SUB CODE: CH

NO REF SOV: 003

OTHER: 000

Card 4/4

ACCESSION NR: AP4018054

AUTHOR: Shikhiyev, I. A.; Aslanov, I. A.

TITLE: Research in the area of synthesis and conversion of unsaturated organogermanium compounds

SOURCE: Zhurnal obshohey khimii, v. 34, no. 2, 1964, 397-398

TOPIC TAGS: Iotsich reagent, organogermanium compound, unsaturated compound, synthesis, conversion, ethylgermanium derivative, diethylgermanium derivative, methyl germanium derivative, diethylgermanium derivative, methyl germanium derivative

ABSTRACT: The reaction of the Iotsich reagent (dimagnesiumdibromine-dimethylethynylcarbinol, dimagnesiumdibrominemethylethylethynylcarbinol and dimagnesiumdibrominediethylethynylcarbinol) with different alkyl- and dialkyl-germanium halides was studied. It was established that the reaction proceeds in the direction of forming corresponding di- and trihydric Y-germaniumacetylene alcohols as follows:

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CESSION	NR: AP4018054	ماندان بيان المساور 				
į.	(CH ₃) ₅ C ₁ C _{GEE} CMgBr	(C,H _s),GeCl _s				
	→ [(CH³)*COHC≡C]					
	→ [(CH ₃) ₈ COHC≡C] OMgBr CH ₃ (C ₈ H ₈)C C≡CMgB	CH ₄ GeBr ₄				
	→ [CH ₃ (C ₈ H ₈)COH	•				
	(CaHth) COME BE					
Card2/3	→ [(C³H²)³COHC	#Alforon3	•		 · · · · · · · · · · · · · · · · · · ·	

ACCESSION NR: AP4018054

Bis(3-methylbutine-1-ol-3)-1-diethylgermanium is described and determined. Four representative germaniumacetylene trihydric alcohols are derived and determined for the first time: tris(6-methylbutine-1-ol-3)-1-ethylgermanium; tris(3-methylpentine-1-ol-3)-1-ethylgermanium; and tris(3-ethylpentine-1-ol-3)-1-methylgermanium. The presence of hydroxyl groups in the composition of ditertiary alcohol is shown by dehydration. The corresponding divinylacetylene organogermanium hydrocarbon bis(2-methylbutine-1-ine-3)-4-diethylgermanium is obtained. Orig. art. has: 1 table.

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Petrochemical Processes, Academy of Sciences, Azerbaidshan SSR)

SUBMITTED: 19Dec62

DATE ACQ: 19Mar64

ENCL: 00

SUB CODE: CH

NO REF SOV: 003

OTHER: 000

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s/0079/64/034/002/0399/0400

ACCESSION NR: AP4018055

AUTHOR: Shikhiyev, I. A.; Garayeva, Sh. V.; Aliyev, M. I.

TITLE: Hydrolytic stability of phenylphenoxy- and phenyl(biphenyloxy)

silanes

SOURCE: Zhurnal obshchey khimil, v. 34, no. 2, 1964, 399-400

TOPIC TAGS: hydrolytic stability, tetrasilane, phenyltrisilane, diphenyldisilane, triphenylphenoxysilane, triphenylsilane, triphenyl, silane, biphenyloxy

ABSTRACT: The hydrolytic stability of (C6H50)4Si, C6H4Si(OC6H5)3, $(c_{6}^{H_{5}})_{2}^{Si}(oc_{6}^{H_{5}})_{2}$, $(c_{6}^{H_{5}})_{3}^{Sioc_{6}^{H_{5}}}$, $c_{6}^{H_{5}^{Si}}(oc_{6}^{H_{4}^{C}}c_{6}^{H_{5}^{-n}})_{2}$, $(c_6H_5)_3^{\text{Sioc}}_6H_4c_6H_5^{-n}, (c_6H_5)_3^{\text{Si}}_6(c_6H_5c_6H_5^{-o})_3, (c_6H_5)_2^{\text{Si}}_6(c_6H_4c_6H_5^{-o})_2,$ and (C6H5)3SiOC6H4C6H5-0 was studied under different conditions (in ether, water, sodium hydroxide solution, in moist air). Regardless

Card 1/2

ACCESSION NR: AP4018055

of the character of the hydrolyzing solution, a sharp variation in hydrolytic stability is observed. This is caused by the number of phenyl groups in the silicon atom. Hydrolytic stability increased in the series $(C_6H_50)_4\text{Si} < (C_6H_5\text{Si}(OR)_3 < (C_6H_5)_2\text{Si}(OR)_2 < (C_6H_5)_3\text{SiOR}$ $(R=C_6H_5, C_6H_5-C_6H_4-n, C_6H_5C_6H_4-n)$. Hydrolytic stability of phenyl (aroxy)silanes depends on the character of the aroxy group and decreases in the series $C_6H_5 > C_6H_5C_6H_5-o > C_6H_5C_6H_4-n$. Hydrolytic stability of phenyl(aroxy)silanes in an alkali medium is considerably lower than in an acid medium. Orig. art. has: 8 figures, 5

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ASSOCIATION: none

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Card 2/2

L 12977-65 EWT(m)/EPF(c)/EWP(j)/T Pc-4/Pr-4 RPL RM ACCESSION NR: AP4037058 8/0079/64/034/005/1393/1395

AUTHOR: Sady*kh-zade, S. I.; Shikhiyev, I. A.; Khalilova, E. M.

TITIE: The addition of silane hydrides to acetylenic alcohols and their derivatives.

SOURCE: Zhurnal obshchey khimi1, v. 34, no. 5, 1964, 1393-1395

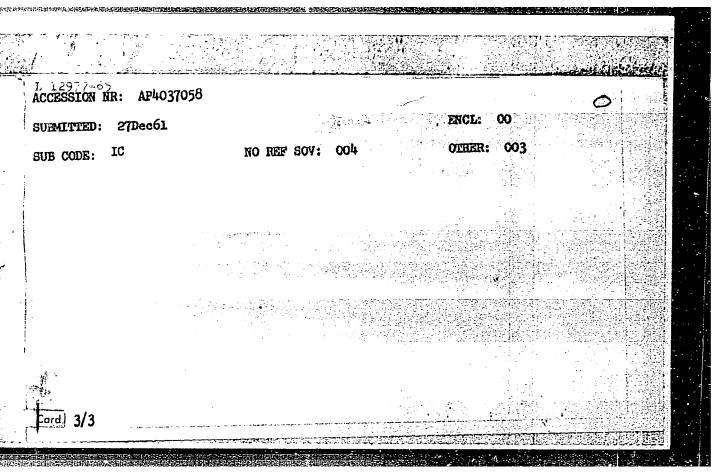
TOPIC TAGS: addition reaction, silane hydride addition product, acetylenic alcohol addition product, propargyl addition product, butyndiol silane hydride adduct, cyanoethylation, triple bond addition reaction.

ABSTRACT: The addition of silane hydrides to propargyl alcohol, butymdiol and propargyl bromide was effected for the first time. It was established that the silane hydrides in the presence of O.IN chloroplatinic acid add to the propargyl alcohol and butymdiol only at the triple bond:

 $\begin{array}{c} \mathrm{CH_3(C_2H_8)_2SIII} + \mathrm{CH} \equiv \mathrm{C} - \mathrm{CH_2OH} \longrightarrow (\mathrm{CH_3})(\mathrm{C_2H_8)_2SICH} = \mathrm{CH} - \mathrm{CH_2OH} \\ \mathrm{R_3SIII} + \mathrm{HOCH_1C} \equiv \mathrm{CCH_2OH} \longrightarrow \mathrm{HOCH_2C} = \mathrm{CH} - \mathrm{CH_2OH} \\ & \\ \mathrm{SIR_3} \\ \mathrm{(II)} \ \ R_3 = \mathrm{CH_4(G_3H_3)_3} \\ \mathrm{(II)} \ \ R_3 = \mathrm{CH_4(G_3H_3)_3} \end{array}$

Card 1/3

L 12977-65	그는 그는 그는 그는 그를 가는 것이 되었다. 그는 그를 가는 것이 되었다. 그는 그를 가는 것이 되었다. 그를 가는 것이 되었다면 되었다. 그를 가는 것이 되었다면 되었다. 그를 가는 것이 되었다면 되었다면 되었다면 되었다면 되었다면 되었다면 되었다면 되었다면		
ACCESSION NR: AP40370)58		
			3
1-methyldiethylsilylpr	open-1-ol-3 (I), 2-methyldiethylbu	ten-2-dio1-1,4 (II) and 17
Same official analysis of the contract of the	Duten-2-0101-1.4 [][] upra premor	mar TT bes T box	[E
1,4-dicyanopropoxybute	iethylsilyl-3-cyanopropoxypropen-l n-2, respectively. Reaction of si	and 2-methyldieth	ylsilyl-
bromide goes in two di	rections:	rana alarades Alfu	propargyl.
	çu₂		
	(RaSIC-CHaBr) → RaSiBr ++ C	H-C-CH.	
R ₃ SIH + HC≡C-	-CH ₂ Br —		
	R ₂ SiCH=CH-CH ₂ Br	(2)	
	$ \begin{array}{c} (VI) R_1 = (G_1H_1)I_1 \\ (VII) R_1 = (GH_2)(G_2H_1)I_2 \end{array} $		
3-bmma-1-trinsonv1a41			
obtained by this react	ylpropen-1 and 3-bromo-1-methyldic ion. Orig. art. has: 3 sets of e	thylsilylpropen-l v	rere
	1	~	
ASSOCIATION: Institut	neftekhimicheskikh protsessov AN A	zerbaydzhanskoy 69	R
THE CITAGE OF LECTOCUE	mical Processes Academy of Sciences	Azerbaydzhanskoy	85R)
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L 62774-65 ENT(m) JAJ/ ACCESSION NR: AP5013768	The same of the sa	.6/65/000/001/0050	/0052	
AUTHOR: Shikhiyev, I. A.;	Abdullayev, N. D.; Akhundova, G. Y			
	-silicon containing acetylene alcoh		3	i I=s
SOURCE: Azerbaydzhanskiy k	chimicheskiy zhurnal, no. 1, 1965,	50-52 B		
TOPIC TAGS: acetylene alco	ohol, gamma silicon acetylene alcoho synthesis, synthesis, alcohol	and the second s		
ABSTRACT: This is the sixt made by the authors since l tertiary γ-silicon acetyle present study concerns the described in the technical termined by obtaining the c	ch study of <u>organosilic</u> on acetylene 1959. Previous studies cover primar one alcohol and some of their chemic synthesis of 4 new silicon acetylene literature. The structu of one of orresponding formal (dimethyloxymet	ry, secondary, and cal reactions. The alcohols not you the alcohols is	d ne et e de-	
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•	িয়া । ১৮ জিলা চালা জিলা জিলা কৰিবলৈ আছিল কৰিবলৈ জিলাইছিল। ১৮ জিলা জিলা জিলা জিলা জিলা জিলা জিলা জিলা		
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ACCESSION NR: AP5013768			
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Card 2/4			

L 62774-65 ACCESSION		013768				ENCTOS	URB: 01	7
Table 1. alcohols	Physicoc	hemical constants of s	iome γ -sil	icon co	ntainin		400000000000000000000000000000000000000	100
	÷ . • - ₆ :						a.	
	Com- pound No.		В.Р. °С/мм	d20	n ₄ ²⁰			
į		I-TriethyIsilyI-3- propin-1-ol l-DimethylethylsilyI- propin-1-ol	3- 100-102/4 78- 80/8	0,8923 0,8740	1,4633 1,4462			
	111	l-Dimethylethylsilyl- methyl-3-butyne-1-ol	3- '8- 89/6	0,8570	1,4402			
!	· iV	l-Diethylpropylsilyl- cethyl-3-pentyne-Lol	3- L 98—100/4	0,8491	1,4514	a de Perfection de la companya de l La companya de la companya de		
	v	n-Butyl-1-triethyl- silylpropyl-1-formal	91- 94/1	0,8834	1,4492			
	<u> </u>					्राचीत्र स्टब्स स्टूर्व । इ.स.च्या स्टब्स स्टूर्व ।	- 1847±255-∓434 *}- 2	

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ACCESSION NR: AP5	013768							• •	ENC	LOSUR	3: 02	
Table 1(Cont.)	, es en										6	
	deter- mined	cal-	Dete:	mine H	đ, ½ Si	General formula	@alc	u late H	51, 51			
	52.79			11,23	16,12	C _e H ₁₈ OS1	63,46	10,65	16,49			
	43,61					C ₁ H ₁₄ OSI	59,09		19.74			
	53,10 71,82		63,76		16,14	C _t C _{ts} OS1 C ₁₃ H ₈₆ OS1	63.46 68,95					
	77,88					C _{I4} H _{I8} O ₄ SI		Įξ				
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L 62806-65 EWT(m)/EPF(c)/EWP(j)/EWA(c) Pc-4/Pr-4 JAJ/RM ACCESSION NR: AP5018354 UR/0316

UR/0316/65/000/002/0059/0063

AUTHOR: Shikhiyey, I. A.; Mekhmandarova, N. T.; Aslanov, I. A.

23

TITLE: Synthesis and conversions of unsaturated organosilicon compounds

SOURCE: Azerbaydzhanskiy khimicheskiy zhurnal, no. 2, 1965, 59-63

TOPIC TAGS: organosilicon compound, organic synthesis

ABSTRACT: This is a continuation of the work with elemental organic derivatives of the secondary acetylenic alcohols [ZhOKh, 33, 377 (1963)]. In this report a synthesis is carried out of some representatives of the γ-silicon-containing secondary alcohols of the ethylens series by reaction of propyl- and isopropylethynyl carbinol with trialkylsilanes in the presence of H₂PtCl₆ according to the following scheme:

R-CHOHC=CH_R'SIH R-CHOHCH=CHSIR's

> $R = (C_5H_7 \cdot uso), C_9H_7 \cdot u, (1-VI)$ $R' = C_9H_8, C_9H_7, C_6H_9.$

Card 1/2

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ACCESSION NR: AP5018354	}			<u>an an Alban an Albanderia (j. 1</u> 1 an Albanderia (j. 11)		
The amount of the						
The presence of a hydrox	yl group	in the	produced alco	hols of the ethyle	ne series	
was proven by formation	or acerd.	rs and	chanostuatro	n. Orig. art. has	: l table.	
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1	EWT(m)/EPF(c)/ENP(j)/T/EWP(t)/EWP(b) IJP(c) JD/RM AP5028891 SOURCE CODE: UR/0316/65/000/004/0042/0043 Shikhiyev, I. A.; Aslanov, I. A.; Mekhmandarova, N. T.; Verdiyeva, S. Sh.
ORG: INK	IP AN AZERSSR JA S
TITLE: S compounds	onthesis and conversions of unsaturated germanium and silicon organic
SOURCE:	zerbaydzhanskiy khimicheskiy zhurnal, no. 4, 1965, 42-43
ethylenic	: organogermanium compound, organosilicon compound, secondary alcohol,
	Sermana arcount
ABSTRACT:	
ABSTRACT:	Ten new compounds were synthesized by reacting propylethynylcarbinol and the trialkylsilanes and trialkylgermanes in the presence
ABSTRACT:	Ten new command

where R = iso-C ₃ H ₇ ; C ₃ H ₇ $R_3^I = CH_3(C_2H_5)_2; CH_3(C_3H_7)_2; CH_3(iso-C_3H_7)_2; C_2H_5(iso-C_3H_7)_2; C_2H_5(C_3H_7)_2.$ The presence of the hydroxyl group in the synthesized compounds was determined by cyanoethylation. Properties of these compounds are tabulated in the original. Orig. art. has: 1 table. SUB CODE: OC/ SUBM DATE: O7Dec64/ ORIG REF: 003/ OTH REF: 001/ ATD PRESS:	. 6485-66 ACC NR: AP50	028891						0
The presence of the hydroxyl group in the synthesized compounds was determined by cyanoethylation. Properties of these compounds are tabulated in the original. Orig. [EW] art. has: 1 table.	here R = iso	o-C ₃ H ₇ ; C ₃ H ₇						
The presence of the hydroxyl group in the synthesized compounds was determined by cyanoethylation. Properties of these compounds are tabulated in the original. Orig. [EW] art. has: 1 table.	های در مها مهمینی و به دیدی در این	سمات اختراد الربيد الربيان الرابية			a # (tao C-1	I_) - • C-H-(Сэна)а.	
eyanoethylation. Properties of these compounds are large representation. [EW]	$R_3^{\bullet} = CH_3$	$(C_2H_5)_2$; $CH_3($	C ₃ H ₇) ₂ ; CH ₃	(iso-C ₃ H ₇) ₂ ;	C2H5(180-C3)	17/2; 02/15\		
eyanoethylation. Properties of these compounds are formula are for	,			. the symthes	ized compour	ds was det	ermined by	
art. has: 1 table.	he presence	of the hydro: ion. Propert:	kyl group 11 ies of these	e compounds a	re tabulated	in the or	iginal. Ori	ig.
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31552-66 ACC NR: AP600	EWT(m)/EWP(j) RM	
APOU	05109 SOURCE CODE: UR/0316/65/000/005/0044/0047	7
AUTHOR: AI	liyev, M. I.; Shikhiyev, I.A.; Balezin, S.A.; Israfilova, S.Z.; Podovayev, N	V. I.
ORG: INKhP	AN Azerb. SSR	80
TITLE: Synti	hesis of unsaturated esters of methylcyclohexylphosphonic acid	78
SOURCE: Az	zerbaydzhanskiy khimicheskiy zhurnal, no. 5, 1965, 44-47	
TOPIC TAGS: Lerivative, US	constant phosphorus compound, organosilicon compound, nonmetallic organite, chemical synthesis, IR analysis, Spectroscopy, corrosion inhibite	nic or
ABSTRACT:	In order to study the reactivity and chemical properties of methylcyclohexyl	l-
pnospnonyl di	chloride, some derivatives were synthesized in accordance with the reaction	n
buosbuonyi di	$CH_3C_0H_{10}POCl_2 \xrightarrow{ROH} CH_3C_6H_{10}PO(OR)_2$	1
phosphonyl di	CHorner, some derivatives were synthesized in accordance with the reaction ROH $CH_3C_0H_{10}POCl_2 \longrightarrow CH_3C_0H_{10}PO(OR)_2,$ $ACC_0H_1OPOCl_2 \longrightarrow CH_2CH_2CH_2CH_2CH_2CH_2CH_2CH_3CH_3CH_3CH_3CH_3CH_3CH_3CH_3CH_3CH_3$	
r,	$CH_3C_0H_{10}POCl_2 \xrightarrow{ROH} CH_3C_6H_{10}PO(OR)_2$	
phosphonyl di	$CH_{3}C_{0}H_{10}POCI_{2} \xrightarrow{ROH} CH_{3}C_{6}H_{10}PO(OR)_{2},$ $ARC R = -CH_{2}-CH = CH_{2}; -CH_{2}-C = CH; -CH_{2}-CH = CH-SI(C_{2}H_{6})_{3};$ $(I) \qquad \qquad (VI) \qquad \qquad -CH-C = CH$	
pnospnonyi di	$CH_3C_6H_{10}POCl_2 \xrightarrow{ROH} CH_3C_6H_{10}PO(OR)_2,$ де $R = -CH_2 - CH = CH_2; -CH_2 - C = CH; -CH_2 - CH = CH - SI(C_2H_6)_3;$ (I) (VI)	

different representat Tylphosphonic acid wer	tives (I-IV) of ethylenic and acetylenic derivatives of re thus synthesized and described for the first time.	methylcyclo-
a triple bond in diprop I also by preparing a s	pargyl methylcylcohexylphosphonate was established s silicon derivative. Infrared analysis showed that the le bond of dipropargyl methylcyclohexylphosphonate f	spectroscopically
e more hydrogenized ca	tion, according to which the radical component, -Si(Carbon atom. A method of synthesis of silicon-phosphorated. Results of a preliminary study of the synthe	C2H5)3, adds to
nowed that the dipropart 30C). Orig. art. has:	gyl ester (II) can be used as a steel corrosion inhibito	esized products or (in 15% HCl at
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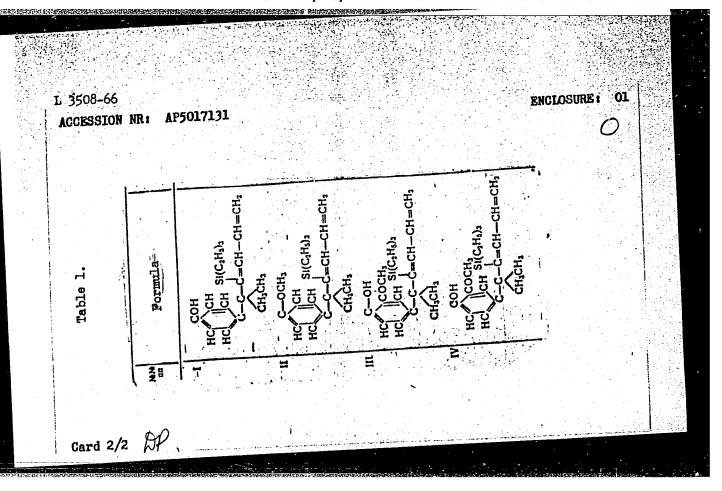
APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001549420004-2"

SHIKHIYEV, I.A.; YUSUFOV, B.G.

Synthesis and transformations of oxygen-containing organic and organosilicon unsaturated compounds. Dokl. AN Azerb. SSR 21 no.1: 19-23 '65. (MIRA 18:5)

1. Institut neftekhimicheskikh protsessov AN AzerSSR.

EWT(m)/EPF(c)/EWP(j)/T L 3508-66 ACCESSION NR: AP5017131 UR/0249/65/021/004/0026/0028 I.; Shikhiyev, I. A.; Salimov, M. A.; TITLE: Synthesis of silicondiene derivatives of aromatic substituted hydrocarbons SOURCE: AN AzerbSSR, Doklady, v. 21, no. 4, 1965, 26-28 TOPIC TAGS: silicon organic compound, triethylsilane, silicondiene derivative ABSTRACT: The reaction between triethylsilane and vinylacetylphenol anisole, gualacol, and o-cresol was investigated. The reaction yielded silicondiene derivatives of benzene substituted hydrocarbons I to IV respectively as shown by Table 1 on the Enclosure. Physical properties and percent yield of the compounds synthesized are tabulated. Orig. art. has: 1 table. ASSOCIATION: Institut meftekhmicheskikh protsessov (Institute for Petrochemical Processes) 44,55 SUBMITTED: 28Jan64 BECL: 01 SUB-CODE: OC, GC NO REF SOV: 007 OTEER: Card 1/2



L 54618-65 ACCESSION NR: AP5008837

Shikhiyev, I. A.; Aslanov, I. A.; Mekhmandarova, N. T.

TITLE: Studies in the field of synthesis and transformations of unsaturated germanium organic compounds. Synthesis and conversions of some monohydric germanium nium organic diacetylene alcohols with isolated triple bonds

SOURCE: Zhurnal obshchey khimii, V. 35, no. 3, 1965, 459-461

TOPIC TAGS: germanium organic compound, organic synthesis

ABSTRACT: The article describes the synthesis of some representatives of germanium organic monohydric diacetylene alcohols with isolated triple bonds using an appropriate Iotsich reagent and germanium organic acetylenic chloride of 5-triethylgermanium-3-methyl-3-chloropent-4-yne according to the reaction $B_{\mathsf{r}}\mathsf{MgOC}(R_{\bullet})C = \mathsf{CMgBr} + (C_{\mathsf{t}}\mathsf{H}_{\bullet})_{\mathsf{s}}\mathsf{GeC} = \mathsf{C}(\mathsf{CH}_{\bullet})(C_{\mathsf{t}}\mathsf{H}_{\bullet})\mathsf{Cl} \\ \rightarrow (C_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})_{\mathsf{s}}\mathsf{GeC} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(C_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} \\ \rightarrow (C_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})_{\mathsf{s}}\mathsf{GeC} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\bullet})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\mathsf{s}})(\mathsf{C}_{\mathsf{t}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{CH}_{\mathsf{s}})(\mathsf{C}_{\mathsf{s}}\mathsf{H}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{C}+\mathsf{C})(\mathsf{C}_{\mathsf{s}})\mathsf{Cl} = \mathsf{C}-\mathsf{C}(\mathsf{C}+\mathsf{C})(\mathsf$ The presence of a hydroxyl group in the obtained compounds was proved through acetylation by reacting diacetylene alcohol with vinylbutyl ether. The article gives physical constants for the synthesized compounds. Orig. art. has: 1 table.

Card 1/2

ACCESSION NR: AP5008837

ASSOCIATION: Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR (Institute of Petrochemical Processes, Academy of Sciences Azerbaydzhan SSR)

SUBMITTED: 24Dec63 ENCL: 00 SUB CODE: OC

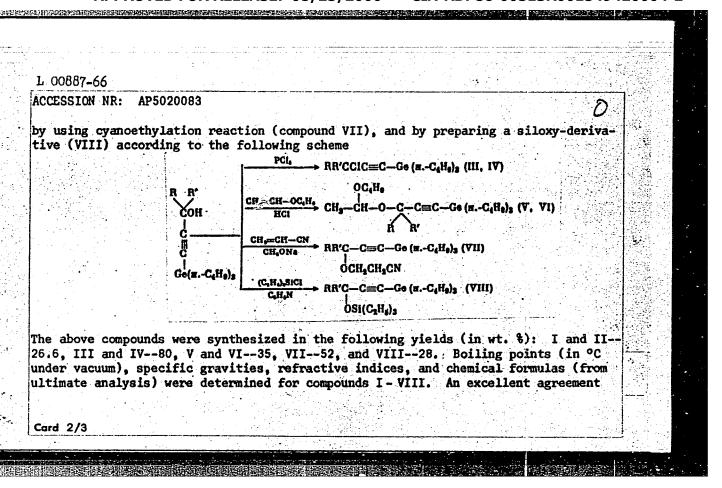
NO REF SOV: 003 OTHER: 000

SHIKHIYEV, I.A.; VATANKHA, A.A.; GUSEYNZADE, B.M.

Synthesis and transformations of oxygen-containing organic and organosilicon compounds. Part 24: Synthesis and transformations of acetylenic formals. Zhur. ob. khim. 35 no.5:812-814 My '65. (MIRA 18:6)

1. Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR.

00887-66 EWT(m)/EPF(c)/EWI	UR/00/5/05/000/
UTHOR: Shikhiyev, I. A.; Abd	
TITLE: Investigations on synt	thesis and reactions of unsaturated organogermanium nd certain reactions of monohydric y-germaniumethyne
olcohol COURCE: Zhurnal obshchey khir	mii, v. 35, no. 8, 1965, 1348-1350
TOPIC TAGS: organogermanium	compound, organomagnesium compound, alcohol, acetylene
ABSTRACT: Reaction of magnes	iumalcoholate of dimethyl- and methylethylethyne carbi- hloride was studied according to the following scheme
RR'C C=CM	AgBr (I, II)
(I) R, R'=CH	on Dack, Regulla
Presence of a hydroxy-group i	in the reaction product (I, II) was proved by preparing V), by using acetalization reaction (compounds V and VI)
Card 1/3	



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SHIKHIYEV, T.A., ALIYEV, M.I., YOSOFOV, B.G.

Studies in the synthesis and transformations of oxygen-containing organic and organosilicon compounds. Part 22; Synthesis of organic and organosilicon trichloroacetals and half ethers of chloro hydrate of the acetylene series. Zhur. ob. khim. 35 no.9:1654-1657 S '65.

(MIRA 18:10)

1. Institut neftekhimicheskikh protsessov AN AzerSSR.

SHIKHIYEV, I.A.; ALIYEV, M.I.; ISRAFILOVA, S.Z.

Synthesis of methyl and dimethylcyclopentylphosphinic acid dichloroanhydrides. Dokl. AN Azerb. SSR 21 no.7:7-9 '65. (MIRA 18:12)

1. Institut neftekhimicheskikh protsessov AN AzSSR. Submitted April 3, 1964.

L 34103-66 EWT(m)/EWP(j) RM ACC NR: AP6008712

SOURCE CODE: UR/0079/65/035/011/2026/2028

AUTHOR: Shikhiyev, I. A.; Abdullayev, N. D.; Aliyev, M. I.; Akhundova, G. Yu.

B

ORG: <u>Institute of Petrochemical Processes</u>, <u>AN Azerbaydzhan SSR</u> (Institut neftekhimi-cheskikh protsessov AN Azerbaydzhanskoy SSR)

TITLE: Studies in the field of synthesis and conversions of unsaturated organosilicon compounds. Part 26: Synthesis of organosilicon formals of the acetylene series

SOURCE: Zhurnal obshchey khimii, v. 35, no. 11, 1965, 2026-2028

TOPIC TAGS: organosilicon compound, acetylene compound

ABSTRACT: The reaction of certain acetylenic organosilicon alcohols with α -chloromethyl alkyl esters in the presence of sodium hydroxide was studied. It was found that the reaction produces the corresponding acetylenic organosilicon formals as follows:

Card 1/2

UDC: 547.314:546.287

L 34103-66

ACC NR: AP6008712

n-Butyl-1-triethylsilylpropyne-1-formal and ten representatives of silicon-containing acetylenic formals were thus synthesized, and their characteristics were determined. Orig. art. has: 1 table.

SUB CODE: 07 / SUBM DATE: 17Oct64 / ORIG REF: 003

Card 2/2 /7/

, ACC NK: AP600952	6 (N) ====	LUP(S) JD/WB/	/ RM /0413/66/000/005/00 4	19/0049	
ATTEROD Bodobou	or W T + Dologin G	A - Chilchivers	T A · Altrew M]		
Israfilova, S. Z	ev, N. I.; Balezin, S.	H. SHIRMLYEV.	Constitution III	76	
ORG: none	1			15.18	
TITLE: Corrosio	n inhibitor for steel	in hydrochloric	acid. Class 22, No	o. 179406.	
[Announced by the	e <u>Moscow State Pedago</u>	gical Institute i	m. V. S. Lenin (Mos	kovskiy	
gosudarstvennyy	pedagicheskiy institut	i))			
SOURCE: Izobret	eniya, promyshlennyye	obraztsy, tovarn	yye znaki, no. 5,	1966, 49	
TOPIC TAGS: ste	el corrosion, corrosio	on inhibitor, aci	d corrosion	10	
				7.4	
ARSTRACT: An Au	thor Certificate has l	een issued for a	corrosion inhibito	or for steel	
in hydrochloric	thor Certificate has bacid. Consisting basi	ically of dipropa	rgyl methylcycloher	cylphosphi-	
in hydrochloric nate. it contain	acid. Consisting basis several other additi	lcally of dipropa lves, preferably	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	
in hydrochloric nate. it contain	acid. Consisting basi	lcally of dipropa lves, preferably	rgyl methylcyclohed in amounts of inhil	cylphosphi-	
in hydrochloric nate. it contain	acid. Consisting basis several other additi	cally of diprope lves, preferably potassium iodide,	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	
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in hydrochloric nate, it contain ~ 0.5%; l-hexyn	acid. Consisting basis several other additi	cally of diprope lves, preferably potassium iodide,	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	
in hydrochloric nate, it contain ~ 0.5%; l-hexyn	acid. Consisting basis several other additi	cally of diprope lves, preferably potassium iodide,	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	
in hydrochloric nate, it contain ~ 0.5%; l-hexyn SUB CODE: 13/	acid. Consisting basis several other additing basis as several other additing basis. 3-01, 0.25 0.5%; I SURM DATE: 07Jan65	cally of dipropa lves, preferably potassium iodide,	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	
in hydrochloric nate, it contain ~ 0.5%; l-hexyn	acid. Consisting basis several other additing basis as several other additing basis. 3-01, 0.25 0.5%; I SURM DATE: 07Jan65	cally of diprope lves, preferably potassium iodide,	rgyl methylcyclohed in amounts of inhil	cylphosphi- pitor BA-6,	

L 45895-66 ENT(m)/ENF(j) ACC NR: AP6026429 (A)	SOURCE CODE: UR/0079/66/036/005/0912/0914
AUTHOR: Shikhiyev, I. A.; Abdu	illayev, N. D.
ORG: Institute of Petrochemica (Institut neftekhimicheskikh pr	l Processes, Academy of Sciences, Azerbaydzhan SSR otesessov Akademii nauk Azerbaydzhanskoy SSR)
TITLE: Synthesis and conversions of	ons of organosilicon compounds. Report No. 30. Synthesilicoacetylenic formals
SOURCE: Zhurnal obshchey khimi	1, v. 36, no. 5, 1966, 912-914
TOPIC TAGS: organosilicon comp	ound, alcohol, ether, acetal
toward , various a-chloromethy	dimary, secondary, and tertiary silicoacetylenic alcohols alkyl ethers was studied. The reactions were:
(Alk) ₂ C—C≡C	$-SiR_3 \xrightarrow{+CiCH_1 - OR'} (Aik)_2 C - C \equiv C - SiR_3$
о́н	OCH ₂ —OR' (I—XI)
(1) R' = (11) R' =	$C_{i}(I_{r}, \dots, R_{i}) = (C_{i}(I_{r})_{i}C_{i}H_{i}, A_{i}K = I_{i}$
(111) R' = (1V) R' = (1V) R' = (V) R' =	$\vec{G}_{1}[i]_{i}, \vec{R}_{1} = (G_{1}[i]_{1})_{i}, A[k = H; G_{1}[i]_{1}]_{i}, A[k = H; G_{2}[i]_{1}]_{i}, A[k = H; G_{3}[i]_{2}]_{i}$
(VI) R' = (VII) R' = (VIII) R' = (VIIII) R' = (VIIII) R' = (VIIII) R' = (VIIIII) R' = (VIIIII) R' = (VIIIIII) R' = (VIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	$C_{1}H_{r}^{-iso}$, $R_{s} = (C_{1}H_{s})_{s}C_{1}H_{s}$, $A_{1}k = C_{1}H_{s}$, $C_{2}H_{3}$
(IX) R' = (X) R' = (X) R' = (X1) R'	
Card 1/2	UDC: 547.367+547.312
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L 45895-66

ACC NR: AP6026429

The structure of the silcoacetylenic formals obtained was determined spectroscopically and by an exchange reaction of one representative with organomagnesium compounds:

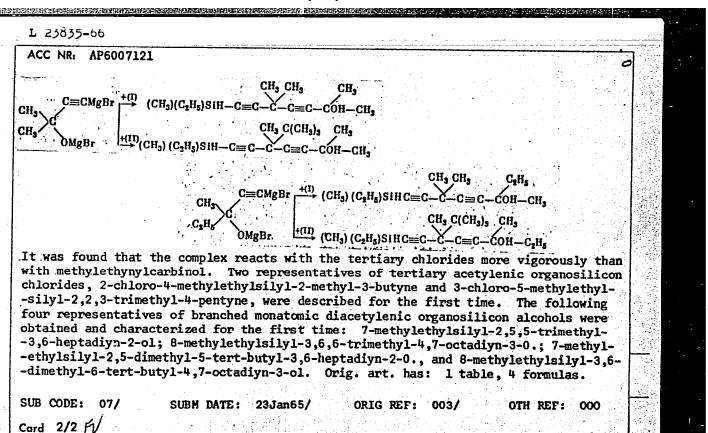
$$\begin{array}{c} \text{CII}_3\\ \text{CII}_3 - \text{C} - \text{C} \equiv \text{C} - \text{Si}(\text{CH}_3)_2 \text{C}_2 \text{H}_5 + \text{Mg} \\ & \begin{array}{c} \text{Br} \\ \text{C}_9 \text{II}_5 \end{array} & \begin{array}{c} \text{C}_9 \text{H}_5 - \text{CH}_2 - \text{OC}_4 \text{H}_6 - \text{R}} \\ \text{CH}_3 \\ \text{OCH}_2 - \text{OC}_4 \text{H}_6 - \text{R} \end{array} & \begin{array}{c} \text{CH}_3 \\ \text{C}_9 \text{II}_5 \end{array} & \begin{array}{c} \text{CH}_3 \\ \text{C}_9 \text{II}_5 \end{array} & \begin{array}{c} \text{CH}_3 \\ \text{C}_9 \text{II}_5 \end{array} & \begin{array}{c} \text{CH}_3 - \text{CC} - \text{C} \equiv \text{C} - \text{Si}(\text{CH}_3)_2 \text{C}_2 \text{H}_3} \\ \text{OH} \end{array} & \begin{array}{c} \text{CH}_3 - \text{CC} - \text{C} \equiv \text{C} - \text{C} = \text{C} - \text{C$$

Eleven representatives were thus obtained for the first time and were characterized. It was found that <u>a-chloromethyl alkyl ethers</u> react more vigorously with tertiary organosilicon alcohols than with primary or secondary alcohols. Orig. art. has: 1 table.

SUB CODE: 07/ SUBM DATE: 03Apr65/ ORIG REF: 001

Card 2/2 LC

ACC NR: AP6007121 SOURCE CODE: UR/0079/66/036/002/0352/0354 AUTHOR: Shikhiyev, I. A.; Rzayeva, S. A.; Guseynzade, B. M. ORG: Institute of Petrochemical Processes, Academy of Sciences, Azerbaydzhan SSR 8 (Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR) TITLE: Studies in the synthesis and conversion of unsaturated organosilicon compounds. Part 27: Branching synthesis of organosilicon alcohols of the biacetylene series SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 352-354 TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound ABSTRACT: The reactions of the magnesium bromide complex 1 CH3 CH3 OMgBr with various tertiary acetylenic organosilicon chlorides were studied. The reactions were as follows:	L 23835-66 EWT(m)/EWP(j)/T RM	
ORG: Institute of Petrochemical Processes, Academy of Sciences, Azerbaydzhan SSR B (Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR) TITLE: Studies in the synthesis and conversion of unsaturated organosilicon compounds. Part 27: Branching synthesis of organosilicon alcohols of the biacetylene series SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 352-354 TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound ABSTRACT: The reactions of the magnesium bromide complex 1 CH3 CH3 OMgBr with various tertiary acetylenic organosilicon chlorides were studied. The reactions	ACC NR: AP6007121 SOURCE CODE: UR/0079/66/036/002/0352/0354	
(Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR) TITLE: Studies in the synthesis and conversion of unsaturated organosilicon compounds. Part 27: Branching synthesis of organosilicon alcohols of the biacetylene series SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 352-354 TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound ABSTRACT: The reactions of the magnesium bromide complex 1 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH	AUTHOR: Shikhiyev, I. A.; Rzayeva, S. A.; Guseynzade, B. M.	
Part 27: Branching synthesis of organosilicon alcohols of the biacetylene series SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 352-354 TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound ABSTRACT: The reactions of the magnesium bromide complex 1 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH	ORG: Institute of Petrochemical Processes, Academy of Sciences, Azerbaydzhan SSR & (Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR)	
TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound ABSTRACT: The reactions of the magnesium bromide complex (CECMgBr CH3	TITLE: Studies in the synthesis and conversion of unsaturated organosilicon compounds Part 27: Branching synthesis of organosilicon alcohols of the biacetylene series	
ABSTRACT: The reactions of the magnesium bromide complex 1 CH ₃	SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 352-354	
CH ₃ C≡CMgBr CH ₃ CH	TOPIC TAGS: organosilicon compound, alcohol, chloride, organomagnesium compound	
with various tertiary acetylenic organosilicon chlorides were studied. The reactions	CECHORD	
	CH ₃ OMgBr	
	with various tertiary acetylenic organosilicon chlorides were studied. The reactions	
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Card 1/2	Card 1/2	



L 23842-66 EWT(m)/EWP(1)/T IJP(c) WW/JW/RM ACC NR: AP6007122 SOURCE CODE: UR/0079/66/036/002/0355/0357 AUTHOR: Shikhiyev, I. A.; Aslanov, I. A.; Verdiyeva, S. Sh. ORG: Institute of Petrochemical Processes, Academy of Sciences, Azerbaydzhan SSR (Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR) TITLE: Studies in the synthesis and conversion of unsaturated organosilicon compounds. Part 28: Synthesis and conversion in certain monatomic organosilicon alcohols of the ethylene series SOURCE: Zhurnal obshchey khimii, v. 36, no. 2, 1966, 355-357 TOPIC TAGS: organosilicon compound, alcohol, chemical naccion ABSTRACT: In order to further develop the studies of organometallic derivatives of unsaturated alcohols, the reaction of isopropylethynylcarbinol with various trialkyliso -C₃H₇CHOHC≡CH H₁PICI₀ iso -C₃H₇CHOHCH=CHSiR₃ R₀ == CH₂(C₂H₂)₂ (I), CH₂(C₂H₁)₃ (II), CH₂(C₄H₂)₂ (III), (C₂H₄)₃ (IV), (C₄H₄)₃ (V). was investigated. Five representatives of secondary organosilicon alcohols of the ethylene series were obtained and described for the first time: 1-methyldiethylsilyl-**Card 1/2**

L 23842-66	<u> </u>				2	
CC NR: AP6007122		ı itamonyleily	1-4-methyl-1-	penten-3-ol; 1	methyldi-	
CC NR: AP6007122 4-methyl-l-penter utyl-silyl-4-meth	n-3-01; 1-methy.	ol: 1-triethy	lsilyl-4-meth	yl-1-penten-3-	the composi-	
4-methyl-l-penter utyl-silyl-4-meth butyl-silyl-4-meth ion of secondary	thvl-l-penten-3	-ol. The pre	sence of hydr	vanoethylation	according to	
ion of secondary	C CITY TOUTE		·		7	
he reaction		CH =CHCN	C H-CH-CH=C	HSIR,		
180-	С3Н7СНОНСН—СИЅІ	R ₃	OCH ₂ —CH ₂	CN.		
100			(IA)			
The corresponding		rivatives of	secondary al	cohols were 'ob'	Catued and de	
the companing	" GASDOSCITORA OF		A Farmulad.	I table.		 (i) (ii) (iii) (iii) (iii) (iii) (iii) (iii) <l< th=""></l<>
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L 04550-67 EWT (ACC NR: AP6025991 $EWT(m) \neq EWP(j)$ SOURCE CODE: UR/0079/66/036/007/1293/1295 AUTHOR: Shikhiyev, I. A.; Vatankha, A. A.; Guseyn-zade, B. M. ORG: Institute of Petrochemical Processes, Academy of Sciences Azerbaydzhan SSR (Institut neftekhimicheskikh protsessov Akademii nauk Azerbaydzhanskoy SSR) TITLE: Synthesis of acetylenic formals and their reactions with hydrosilanes SOURCE: Zhurnal obshchey khimii, v. 36, no. 7, 1966, 1293-1295 TOPIC TAGS: organosilicon compound, organic synthesis ABSTRACT: The purpose of this investigation was to compare the reactivity of certain acetylenic hydrosilane derivatives. Synthesis of acetylenic cyclohexylformals and their reactions with hydrosilanes were conducted by the following scheme: $-C \equiv CH \xrightarrow{\text{CiGH}_1\text{OR}} ROCH_2O \cdot C - C \equiv CH + NaCl + H_2O$ $R = CH_{3}, C_{3}H_{4}, C_{2}H_{7}H., C_{3}H_{7}H80, C_{4}H_{4}$ HSIR, ROCH20 C/ ROCH₂O C C CH H₁PtCl₀ $R = CH_{2}, C_{1}H_{T}H30, C_{4}H_{1}, (OC_{1}H_{1})_{1}, CH_{3}(C_{1}H_{1})_{1}, CH_{3}(C_{3}H_{1})_{2}.$ UDC: 547.362+547.245 Card 1/3

